

Piranha gobbles up hysteresis of aluminium oxide–nitride semiconductor interface

The Naval Research Laboratory shows how nitride surface pre-treatment can cut charge traps to almost a quarter for Al₂O₃ gate insulator.

US Naval Research Laboratory researchers (with Nelson Y. Garces of Global Strategies Group) have been studying ways to improve the quality of aluminium oxide (Al₂O₃) grown on gallium nitride (GaN) by atomic layer deposition (ALD) [Neeraj Nepal et al, *Appl. Phys. Express*, vol4, p055802, 2011]

Aluminium oxide is of interest as a gate insulator for high-frequency, or high-power, nitride semiconductor transistors to reduce gate leakage currents compared with devices that use a direct Schottky metal gate contact with the semiconductor barrier. Other advantages of an insulated gate are higher breakdown voltages and higher channel current.

However, producing a high-quality insulator–semiconductor interface is tricky. In particular, one needs to avoid charge traps and fixed charges from developing that adversely affect device performance. For this, some surface preparation of the semiconductor surface before deposition is needed. For convenience, it is preferable to perform this preparation in the deposition chamber itself, rather than having to use an external process.

A popular technique for creating Al₂O₃ films is atomic layer deposition. Surface preparation in such chambers requires a relatively low-temperature process.

Three pre-treatments were studied: hydrogen peroxide–sulfuric acid (H₂O₂:H₂SO₄, 1:5), hydrofluoric acid solution (H₂O:HF, 1:1), and hydrochloric acid solution (H₂O:HCl, 1:1). Hydrochloric acid is commonly used as a native oxide etchant for GaN (e.g. for removal of oxides that form on exposure to air). The acid solution treatments were carried out for one minute at room temperature. The hydrogen peroxide–sulfuric ‘piranha’ treatment lasted for 10 minutes at 80°C. The substrate consisted of 2µm of n-GaN on sapphire.

The atomic layer depositions consisted of 250 cycles of alternating 15ms pulses of deionized water and trimethyl-aluminium (TMA), separated by 20s purges to remove unreacted precursors. The carrier gas was nitrogen. The deposition temperature was either 240°C or 260°C. The resulting film was measured to be 22nm.

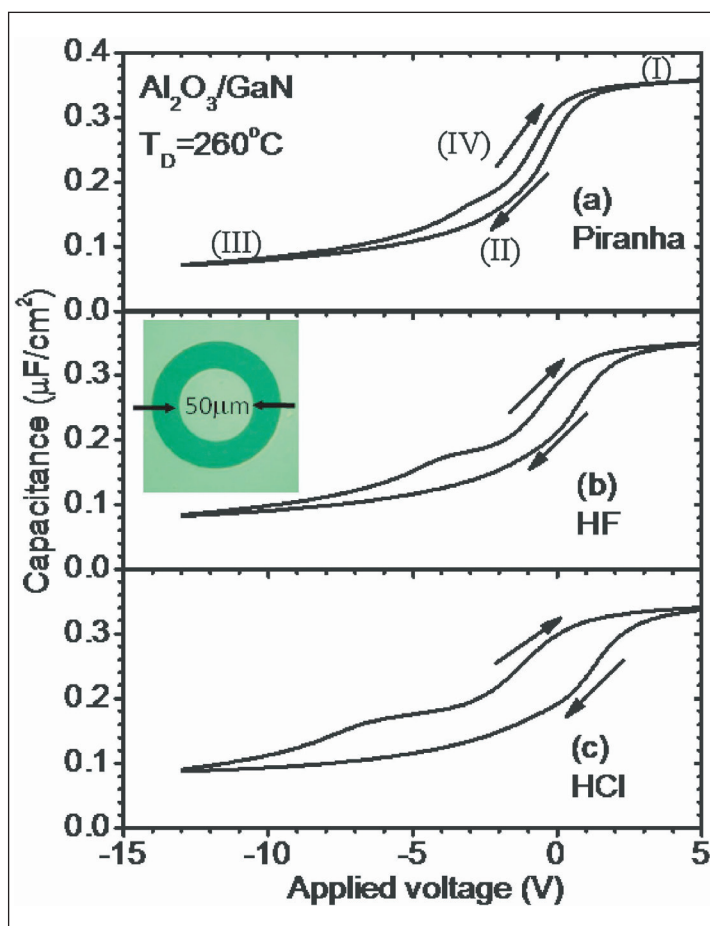


Figure 1. Capacitance–voltage curves of Al₂O₃ on n-GaN for (a) piranha, (b) HF, and (c) HCl surface treatments at deposition temperature of 260°C. Inset of (b) shows microscope image of 50µm diameter capacitor.

Circular metal-oxide-semiconductor capacitor structures were formed using photolithography and electron-beam evaporation/liftoff of a nickel-gold metal stack. The top contact had a diameter of 50µm, separated from a concentric large-area 3.4×10⁷µm² contact by a gap of 20µm.

The piranha and hydrofluoric treatments resulted in uniform and smooth layers of Al₂O₃, but the hydrochloric treatment gave non-uniform nucleation with 10–30nm diameter particles being shown in

atomic force microscopy (AFM) examinations. Such non-uniform nucleation can generate point defects (vacancies, vacancy-complexes, and interstitials) that potentially lead to higher densities of charge traps and fixed charges.

Capacitance-voltage measurements (Figure 1) showed a larger hysteresis for the sample pre-treated with hydrochloric acid, confirming that the non-uniform nucleation did indeed result in higher trap and fixed charge densities.

The researchers calculated a dielectric constant for their Al_2O_3 film of ~ 8.9 (with piranha pre-treatment). The behavior at the flat-band voltage was used to quantify the amount of hysteresis (Table 1). The measurements suggest the presence of negative trapped charge at or near the $\text{Al}_2\text{O}_3/\text{GaN}$ interface. On this measure, the piranha treatment at 260°C more than halved the trapped charge compared with hydrofluoric treatment, and almost quartered it compared with hydrochloric treatment. A similar trend was seen with 240°C treatment.

Although there are a number of possible explanations for the hysteresis, such as oxide trapped charge, interface trapped charge, and/or mobile ionized impurities, the researchers suggest that there are oxide and interface trapping states at or near the $\text{Al}_2\text{O}_3/\text{GaN}$ interface that are being charged and discharged (Figure 2). X-ray photoelectron spectroscopy did not find impurity-related peaks, leading the researchers to say that it is 'unlikely that ionized impurities contribute significantly'.

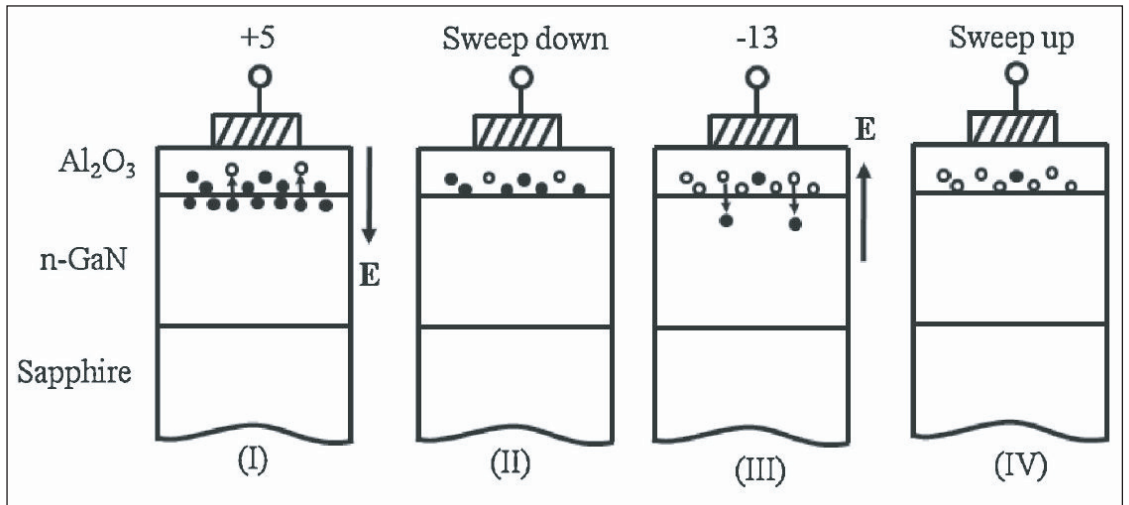


Figure 2. Schematic of electrons and their motion near the $\text{Al}_2\text{O}_3/\text{GaN}$ interface due to applied bias at the (I) accumulation, (II) positive to negative bias polarity sweep (sweep down), (III) depletion, and (IV) negative to positive bias sweep (sweep up) regions. Corresponding hysteresis curve positions are marked in Figure 1(a). Closed circles represent electrons and open circles represent neutral or positive trapping states.

The piranha treatment at 260°C more than halved the trapped charge compared with hydrofluoric treatment, and almost quartered it compared with hydrochloric treatment. A similar trend was seen with 240°C treatment

The researchers comment: 'During (I) accumulation, we suspect that electrons are injected into or captured by nearby trapping states and held there during (II) the sweep from accumulation to depletion, causing a positive V_{FB} shift. In (III) deep depletion, we suspect that the large electric field causes Poole-Frenkel or direct-tunneling electron emission from oxide and interface trapping states, resulting in a net positive shift in trapped charge. Since trap states are no longer occupied during (IV) the sweep from depletion to accumulation, V_{FB} shifts negatively until electron injection/capture can take place again during (I) accumulation.'

On the basis of the present results, it is difficult to determine the relative contributions of oxide and surface state contributions to the trapping. However, a 'surface-state ledge' is seen in the hydrochloric treated sample (Figure 1(c)) that diminishes progressively with the hydrofluoric and piranha treatments. ■ <http://apex.jsap.jp/link?APEX/4/055802>
The author Mike Cooke is a freelance technology journalist who has worked in the semiconductor and advanced technology sectors since 1997.

Table 1. Summary of oxide and flat band capacitance, dielectric constant, hysteresis at V_{FB} , and total trapped charge density of ALD Al_2O_3 deposited at 260°C on n-GaN for different surface pre-treatments.

Pre-treatment	C_{OX} ($\mu\text{F}/\text{cm}^2$)	C_{FB} ($\mu\text{F}/\text{cm}^2$)	ϵ	Hysteresis at V_{FB} (V)	Total trapped charge ($\times 10^{12}/\text{cm}^2$)
Piranha	0.357	0.324	8.9	0.54	1.2
HF:H ₂ O (1:1)	0.348	0.317	8.7	1.15	2.5
HCl:H ₂ O (1:1)	0.340	0.309	8.4	2.13	4.5